#### 80 ON FIXED AIR,

skin, and some degree of drowsiness, but evidently abates the serous discharge, and diminishes the eruption. This patient has several symptoms which indicate a genuine scorbutic diathesis; and it is probable that fixed air taken internally would be a useful medicine in his case.

The faline draughts of Riverius are supposed to owe their antiemetic effects to the air, which is separated from the salt of wormwood, during the act of effervescence. And the tonic powers of many mineral waters seem to depend on the same principle. But I shall exceed my design by enlarging further on this subject. What has been advanced, it is hoped, will suffice to excite the attention of Physicians to a remedy, which is capable of being applied to so many important medicinal purposes.

ON THE ANTISEPTIC AND SWEETEN-ING POWERS, AND ON THE VARIE-TIES OF FACTITIOUS AIR.

HOUGH the fact has lately been controverted by an ingenious writer, I am fully convinced with Dr. Macbride, from the evidence of repeated experiments, that fixed air has the property both of retarding and of correcting putrefaction. It may afford matter of amusement, to consider in what manner these effects are produced.

THAT fixed air may restrain, and even prevent putrefaction, without possessing any inherent antiseptic quality, is not difficult to conceive. For by surrounding the putrescent substance with that kind of air, which

But

(a) See Doctor Priestley's most ingenious papers on factitious air, which will probably be published in the 62d. vol. of the Philosophical Transactions.

# ON FACTITIOUS AIR. 83

Bur supposing the foregoing hypothesis to be well founded, which I advance only as conjecture, how are we to explain the sweetening powers of fixed air? An eminent philosopher seems to hint that fixed air may act as a menstruum for the putrid effluvium, and thus imbibe or difcharge it from the feptic body. The same idea suggested itself to Mr. Henry, in confequence of the following experiment, to which I was a witness. A piece of putrid flesh was suspended twelve hours, in a three pint bottle closely corked, and filled with fixed air, which had been feparated from chalk by the vitriolic acid. The beef was confiderably sweetened, but the air in the bottle was rendered intolerably offensive. Now it affords a natural folution of this fact, if we admit that fixed air, by the laws of chemical affinity, abstracts from the septic body, and holds fuspended or dissolved the putrid particles which it emits. And fuch an affinity feems probable, from their ready combination. G 2

### 84 ON FACTITIOUS AIR.

bination, as well as from their disposition to fly off together from putrefying substances. But how is the putrefactive process checked, and the fresh generation of effluvia restrained under such circumstances? A piece of the same slesh, which was employed in the foregoing experiment, was left all night in the external air, by the circulation of which the effluvia could not fail to be carried off as they were formed; yet the offensive odour of the flesh was not diminished. Has not the reason of this difference, between the exposure of a putrid substance to common air, and to mephitic air, been before asfigned, when it was suggested that the latter may perhaps restrain the slight of that principle in bodies, the separation of which constitutes an essential part of the process of putrefaction? Animal flesh will neither become putrid in vacuo, nor when closely confined from the access of common air. In both cases a vehicle is wanting for the escape of the mephitic

## ON FACTITIOUS AIR. 85

air. In like manner red hot wood ceases to burn, in inflammable air, because such air is already saturated with phlogiston.

I HAVE advanced the preceding conjectures, concerning the manner in which fixed air may retard and correct putrefaction, not as affording me full conviction, or to indulge a fanciful hypothesis, but to promote the further investigation of a subject so curious and interesting.

#### EXPERIMENT I.

It is a fact lately ascertained by a very accurate Philosopher, that putrefaction generates air similar to that which animals have breathed. But this and the succeeding experiment shews that there is some little diversity in their properties and effects. Air was blown forcibly from the lungs, for a sufficient length of time, into a phial containing distilled water and iron silings. The water was then silter-

 $G_3$ 

ed,

### 86 ON FACTITIOUS AIR.

ed, and a few drops of the infusion of galls were added to it. A dark red colour, inclining to purple, was instantly produced.

#### \*EXPERIMENT II.

Eight ounces of ox gall were poured into a bottle, which had a tube communicating with another phial, containing half an ounce of iron filings, and four ounces of distilled water. After standing two days, part of the water was filtered, and suffered no change of colour from the addition of an astringent tincture. But the next day, when the fermentation in the gall was more evident, another filtered portion of the water struck with the same tincture a deep rosy red. On the fifth and sixth days, when the gall became intolerably putrid, though the vapour still corroded the iron filings, it seemed to have

\* Communicated by Dr. Palconer of Bath.

# ON FACTITIOUS AIR: 87

lost the power of dissolving them. For the astringent tincture no longer produced any change of colour in the water, and the iron was evidently precipitated.

#### \*EXPERIMENT III.

Solutions of iron in water, obtained by different kinds of fixed air, vary in the colours which they strike with an infusion of galls. When the vitriolic acid and fossil alkali are employed, a black tinge is produced; when magnesia, or calcareous earths and the same acid are used, a purple hue is struck; and when the air is supplied by fermentation, the artificial chalybeate is changed by galls into a rosy red.

## EXPERIMENT IV.

AIR discharged from chalk by the vitriolic acid readily and perfectly combines G 4 with

By the fame.

with water; but when separated by the nitrous acid, the union is more difficult to be effected, and much less complete. And the artificial mineral water, made by the latter, is more pungent and sparkling than by the former acid.

## EXPERIMENT V.

FACTITIOUS AIR, separated from steel filings by the vitriolic acid, neither occasioned any precipitation in lime water, nor rendered the caustic fixed alkali mild. Whereas the air set free from chalk and magnesia, by the same acid, instantly produced a milkiness in lime water, and restored to the caustic alkali the power of effervescence.

### EXPERIMENT VI.

A PIECE of putrid mutton, which had been employed as a standard in some or ther experiments, was divided into two equal

# ON FACTITIOUS AIR. 89

equal parts: One of these was suspended by a thread in a phial, containing an effervescing mixture of chalk and dilute spirit of vitriol; the other in a similar phial, with a mixture of iron filings and the same acid. The mouths of the phials were slightly stopped with folded paper; and a brisk fermentation took place in each of them. After being exposed fixteen hours to the air detached from these substances, the bits of mutton were taken out, and examined. They were both confiderably firmer in their texture; and the one which had been suspended over the effervescing mixture of chalk and oil of vitriol was entirely sweetened; but the putrid fetor of the other was not in the least degree corrected.

#### EXPERIMENT VII.

A PIECE of putrid flesh was suspended about half an hour over a mixture of iron filings and nitrous acid, and was perfect-

## 90 ON FACTITIOUS AIR.

ly fweetened. It had acquired a pungent and flightly acid smell, but remained sirm and free from setor when this odour was washed off. The water, in which the slesh was washed, did not effervesce with lixivium tartari; nor did the vapour arising from the spirit of nitre and iron silings produce any change of colour in a paper covered with syrup of violets; presumptive proofs that the sweetness of the slesh was not restored by any acid sumes.

The fixed air of metals seems, by some of these experiments, to be of a kind different from that which is contained in alkalis and calcareous earths. And consequently the action of these substances as fluxes, cannot be explained on the principle of their restoring the air which had been lost by calcination. Indeed there are other proofs that the resuscitation of calces does not depend on this cause. I have been assured by an able Chemist that he has repeatedly restored minium to

### ON FACTITIOUS AIR.

its metalline state, by the caustic alkali, assisted by a proper degree of heat; and that several of the metals may be revived by the force of fire alone. It is true that a mild calcareous earth employed as a flux, is always rendered caustic by the operation. But this may be owing to the action of the fire, and not to the loss of its air by elective attraction. Perhaps the operation of alkalis and calcareous earths as fluxes may depend on their absorbing the matter which feems to be added to metallic substances by the process of calcination, and which furnishes such an amazing increase of weight (a)? Inflammable bodies may produce the same effect, by volatilizing and carrying it off.

(a) Antimony, when calcined, gains one eleventh part of its original weight; zinc one tenth; tin one fixth; and lead, when converted into minium, one fourth.